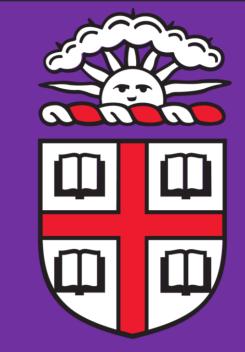
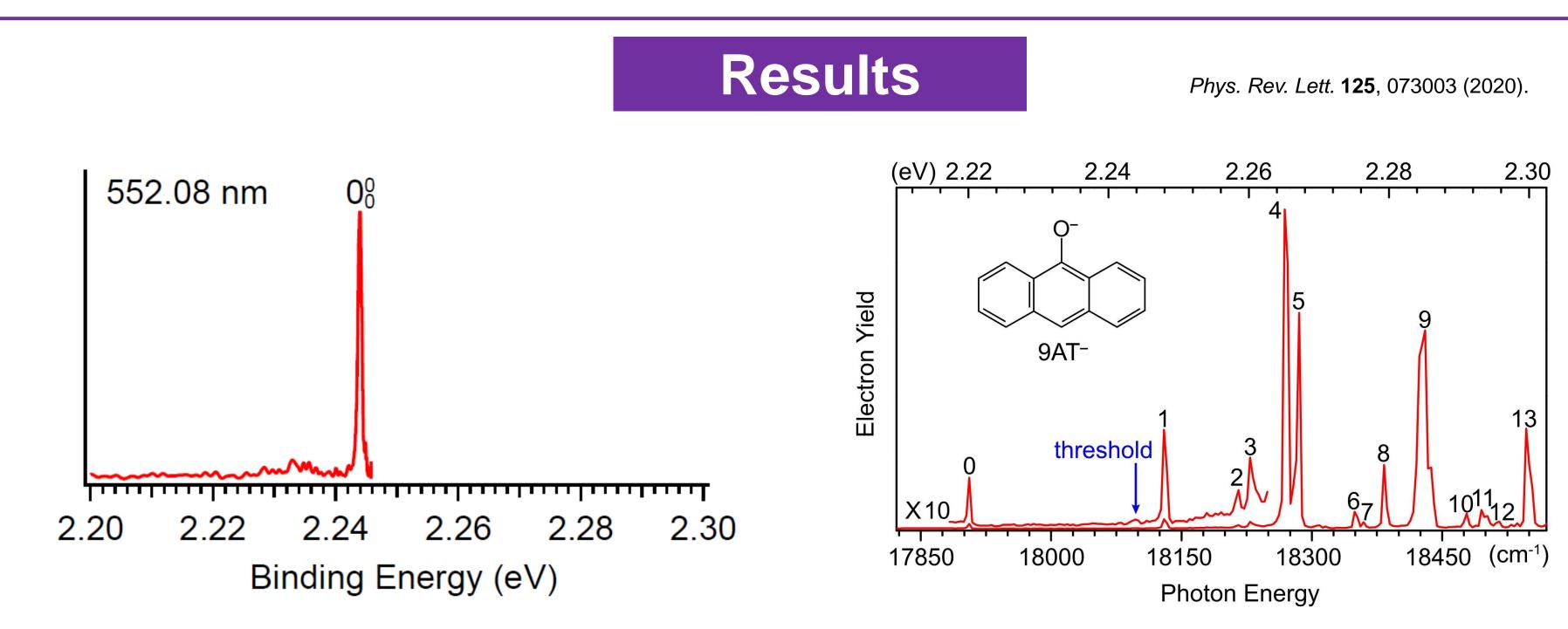
Observation of a π -Type Dipole-Bound State in Molecular Anions Dao-Fu Yuan, Yuan Liu, Chen-Hui Qian, Yue-Rou Zhang, Brenda M. Rubenstein, Lai-Sheng Wang*

Department of Chemistry, Brown University, Providence, Rhode Island 02912, United States



Introduction

Polar molecules with sufficiently large dipole moments can form highly diffuse dipole-bound anions, and possess noncovalent dipole-bound states (DBSs) just below the electron detachment threshold. DBSs have been proposed as the "doorway" to the formation of valence-bound anions, and exploited as a means to develop high resolution resonant photoelectron spectroscopy. The π -type DBS (π -DBS) with one additional angular node in the wave function was predicted to exist as an excited electronic state in addition to the σ -type ground state. However, π -type noncovalent states have never been observed in isolated molecular systems. Here we report the observation of a π -DBS in cryogenically cooled 9-anthrolate anions (9AT⁻, C14H9O⁻) by resonant two-photon photoelectron imaging. A DBS is observed 191 cm⁻¹ (0.0237 eV) below the detachment threshold, and the existence of the π -DBS is revealed by a distinct (s + d)-wave photoelectron angular distribution. The π -DBS is stabilized by the large anisotropic inplane polarizability of 9AT. The population of the dipole-forbidden π -DBS is proposed to be via a nonadiabatic coupling with the dipole-allowed σ -type **DBS mediated by molecular rotations**



Non-resonant photoelectron spectrum of 9AT⁻ at 552.08 nm. The detachment threshold is 2.2437 eV.

The photodetachment spectrum of 9AT-.

1¹2¹

 $4^{1}/5^{1}$

2¹

★ BE_{DBS} = 191 cm⁻¹ (0.0237 eV)

 \Rightarrow EA = 18097 cm⁻¹ (2.2437 eV)

DBS

 $6'^{1}7'^{1}$

 $2^{\prime 1} 11^{\prime 1} / 2^{\prime 1} 4^{\prime 2} / 2^{\prime 1} 5^{\prime 2} / 1^{\prime 1} 2^{\prime 1} 9^{\prime 1}$

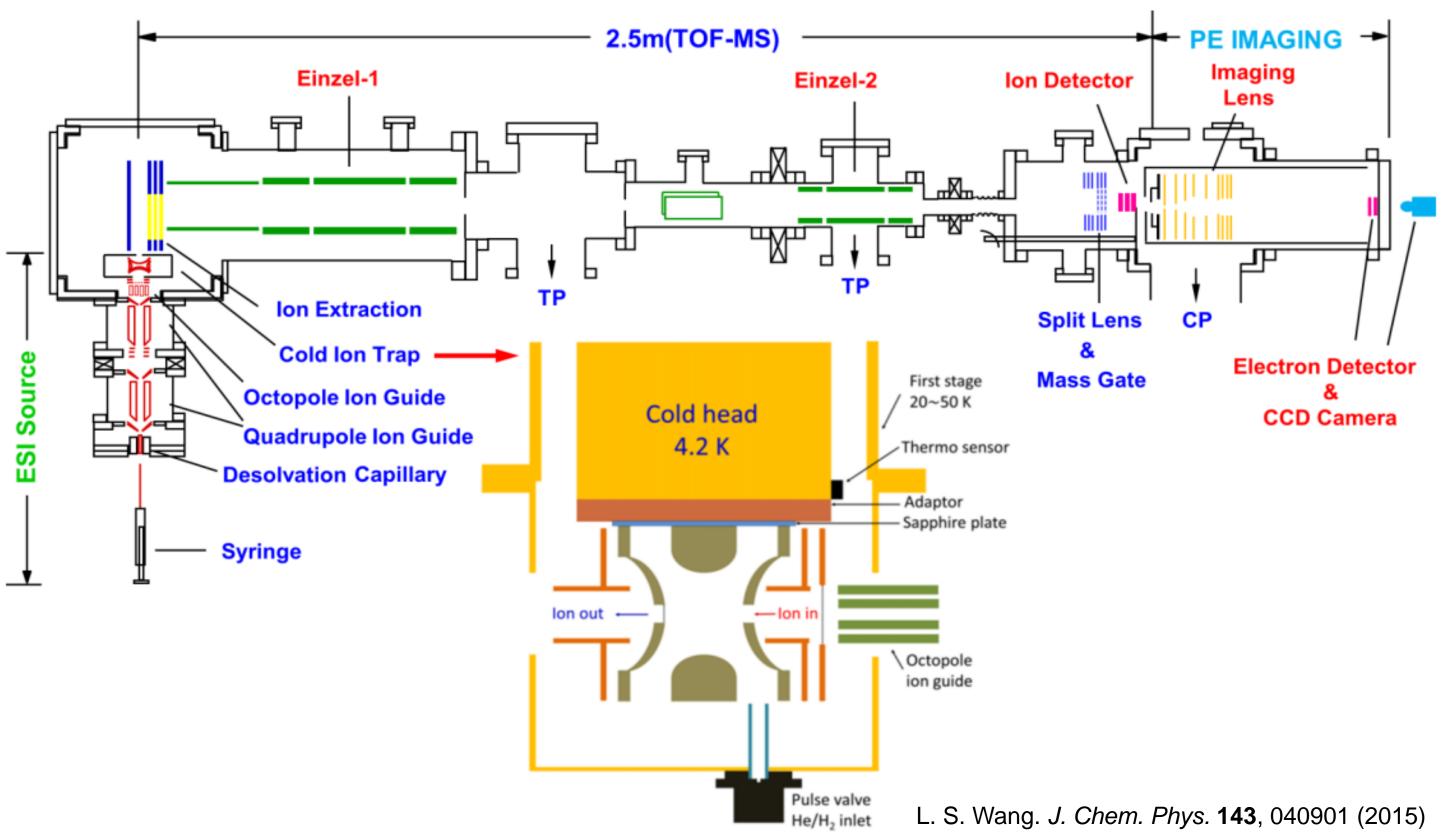
 $4'^2/5'^2$

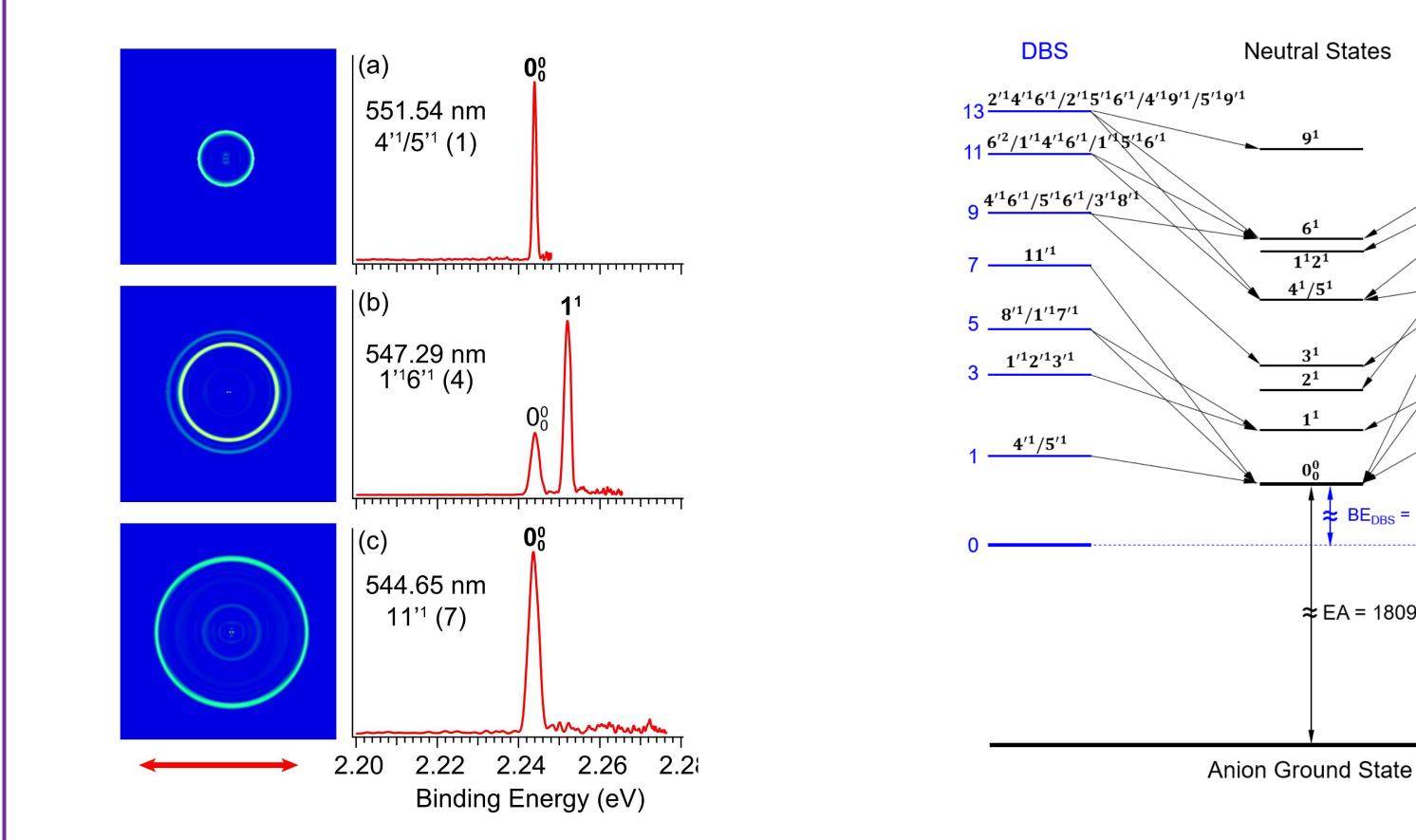
 $1'^{1}6'^{1}$

 $7'^{1}$

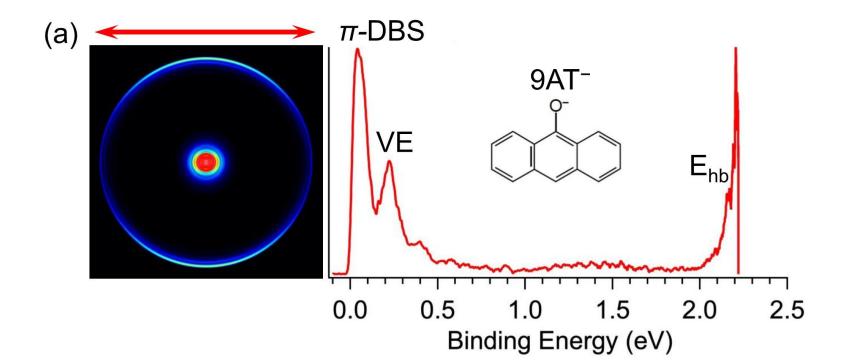
2'13'14'1/2'13'15'1

Experimental Apparatus



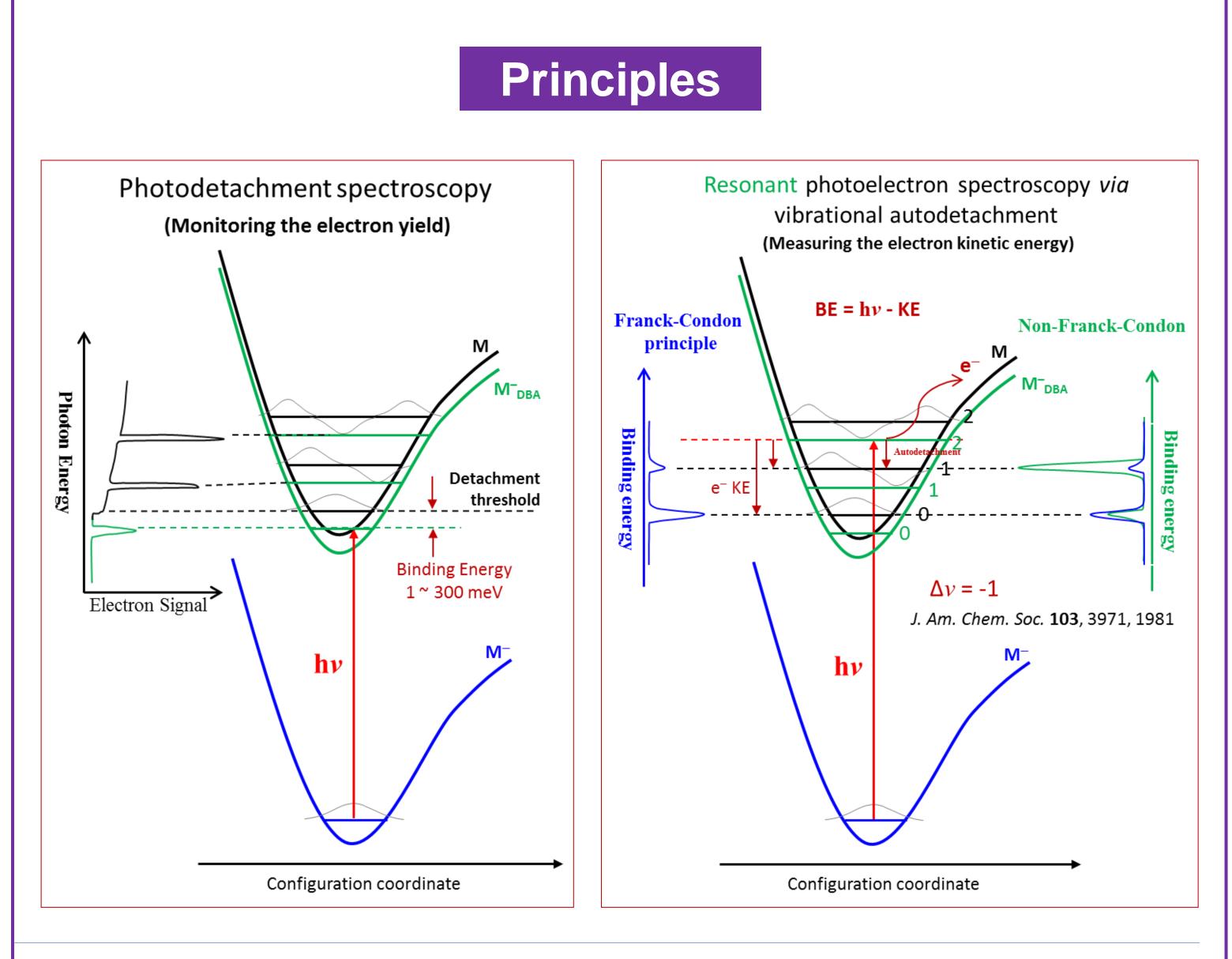


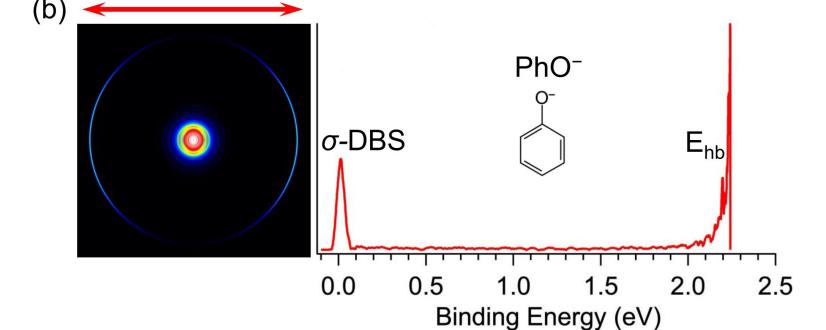
(left) Three Resonant PE images and spectra of 9AT⁻. (right) Schematic energy level diagram for autodetachment from the DBS vibrational levels of 9AT⁻ to the related neutral final states.



Polarizability	9AT	PhO
α_{xx}	78.5	39.2
α_{xy}	0.0	0.0
α_{yy}	284.0	77.1
α_{xz}	0.0	0.0
α_{yz}	0.0	0.0
α_{zz}	173.7	96.9
α_{iso}	178.8	71.1

Electrospray photoelectron spectroscopy apparatus equipped with a cryogenically-cooled Paul trap and a high-resolution photoelectron imaging system

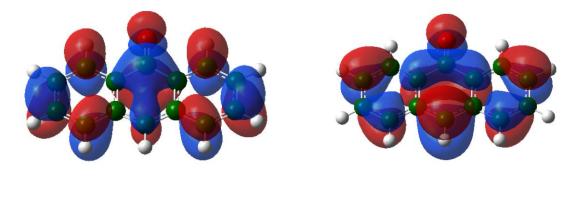




Photoelectron images and spectra of 9AT- and PhOfrom resonant two-photon detachment (R2PD) via the vibrational ground states of their respective DBS

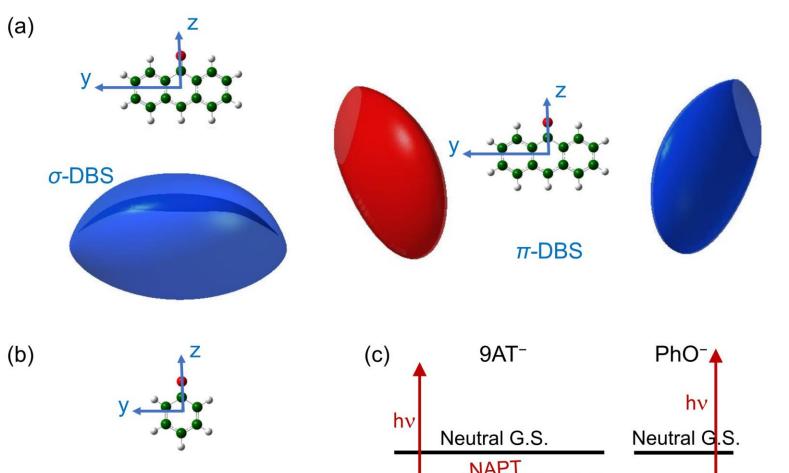
	9/	↓T -
	Cal	Exp
σ-DBS (cm ⁻¹)	336	191
(meV)	41.7	23.7
π-DBS (cm ⁻¹)	261	116
(meV)	32.4	14.4
EA (cm ⁻¹)	16884	18097

The theoretical polarizability tensor α of the neutral 9AT



LUMO

HOMO



Acknowledgement

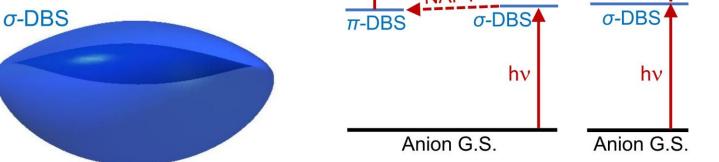
This work was supported by the DOE, Office of Basic Energy Sciences under Grant No. DESC0018679 (to L. S. W.). B. M. R. thanks the Alfred P. Sloan Foundation. Y. L. was supported by the Brown Presidential Fellowship and the Open Graduate Education Fellowship. The calculation was performed using computational resources and services provided by CCV of Brown University.





(eV) 2.0934 2.2437

Binding energies of the DBSs at the CAM-B3LYP/Def2-TZVPP+4s3p2d1f level of theory; The electron affinity (EA) of the neutral 9AT at the B3LYP/Def2-TZVPP level of theory.



The DBS orbitals and a schematic diagram of the R2PD processes



The observation of different weakly-bound states in the same molecule and their sensitivity to molecular structures make it possible to control molecular quantum states via molecular design. The distinct nodal structure of the π -DBS as compared to the σ -DBS also make them candidates as molecular qubits for quantum information science applications. The unique nonadiabatic dynamics mediated by molecular rotation from the σ -DBS to π -DBS further opens up new directions for understanding the electron capture mechanisms by neutral molecules to form molecular anions.