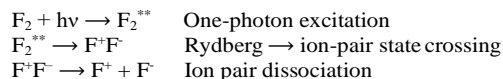


Resolving the F₂ Bond Energy Discrepancy Using Coincidence Ion Pair Production (cipp) Spectroscopy.

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During coincidence ion pair production (cipp) measurements a signal is recorded only when anions and cations are produced simultaneously. The coincident ion production (F⁺ + F⁻) measured in this study is formed via one-photon excitation to bound high energy Rydberg states, followed by state crossing to ion-pair states above the dissociation limit and subsequent dissociation to form F⁺ and F⁻ ions. This mechanism can be presented as:



Cipp spectra of F₂ were recorded on the DELICIOUS III coincidence spectrometer¹ in the one-photon excitation region of 125 975–126 210 cm⁻¹. The F⁺ + F⁻ signal in this frequency range shows rotational band head structure, corresponding to F₂ Rydberg states crossing over to the ion pair production surface.

Through a simulation of the spectra and quantum defect analysis, five Rydberg states spectra were characterized and assigned, four ¹Σ_u states and one ¹Π_u state. (Figure 1).

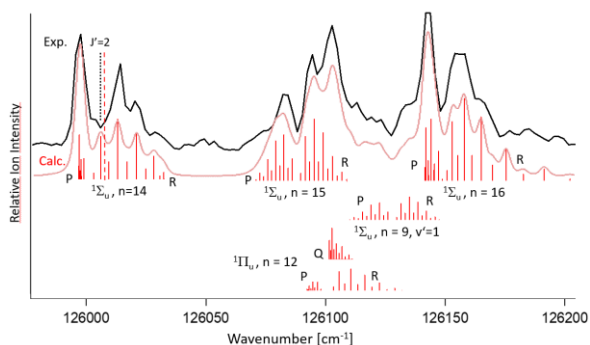


Fig. 1. F₂ coincidence ion pair production spectra in the 125 950 – 126 210 cm⁻¹ photon energy region. Experimental spectrum (black, top), calculated spectrum (red, middle) using 5 cm⁻¹ linewidths, and rotational lines (red sticks, bottom). The ion pair threshold is indicated by a red dashed line. J'=2 for the ¹Σ_u⁺[3/2]14pπ_u state is indicated by a dashed black line. Assignments for the spectral contributions are indicated

As can be seen from the figure some of the assigned states overlap each other. This makes accurate determination of the spectroscopic constants somewhat problematic, resulting in large error bars. The spectroscopic parameters derived from the simulation are listed in Table 1.

Table 1. Rydberg state specifications ($Ry^{2\Sigma^+1X}[Q_c]nl$) for $2\Sigma^+1X$ = term symbol for the Rydberg state) vibrational quantum numbers (v'), band origin (v_0') and rotational parameters (B' , D') based on simulation shown in Fig. 1 and quantum defect analysis.

State assignments	v'	Band origin (v_0') [cm ⁻¹]	$B_e'^a$ [cm ⁻¹]	$D_e'^a$ [cm ⁻¹]
¹ Σ _u ⁺ [3/2]14pπ _u	0	125 999	1.18 ± 0.01	0.0015 ± 0.0003
¹ Σ _u ⁺ [3/2]15pπ _u	0	126 086	0.9 ± 0.1	(0)
¹ Π _u [1/2]12pσ _u	0	126 099	1 ± 0.1	(0.0002)
¹ Σ _u ⁺ [3/2]9pπ _u	1	126 126	0.9 ± 0.1	(0)
¹ Σ _u ⁺ [3/2]16pπ _u	0	126 147	0.96 ± 0.1	(-0.0014)

^aPrecision of parameters is affected by overlap of spectra and rotational line overlaps.

One of the key features of the simulation is the apparent lack of the J' = 2 rotational peak in the first sigma state of the experimental spectrum (at around 126 000 cm⁻¹). This peak, which should occur at 126 005 cm⁻¹ (see Fig. 2) must be missing because its excited state energy level is below the ion pair production threshold.

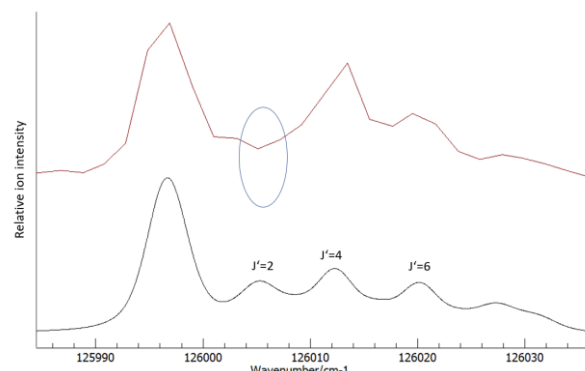


Fig. 2. Comparison of simulation to experimental data showing clear lack of the J'=2 rotational peak belonging to the ¹Σ_u(n=14, v'=0) state.

This finding allows us to estimate, with high certainty, the F₂ → F⁺ + F⁻ dissociation threshold, which can also be used to determine the F–F bond dissociation energy, D₀(F₂). The energies (E) of the rotational levels J' = 2 and J' = 3 of the ¹Σ_u⁺14pπ_u state, converging to the ²Π_g[3/2] ground state of F₂⁺ are,

$$E(^1\Sigma_u^+[3/2]14p\pi_u(v'=0, J'=2)) = 126\,004\text{ cm}^{-1} = 15.6225\text{ eV}$$

$$E(^1\Sigma_u^+[3/2]14p\pi_u(v'=0, J'=3)) = 126\,011\text{ cm}^{-1} = 15.6234\text{ eV}$$

This means that the ion pair production threshold energy for F₂ → F⁺ + F⁻ can be given as 15.6229₄ ± 0.0004₃ eV. Using the F adiabatic ionization energy of 17.42283 ± 0.00005 eV² and its electron affinity of 3.401190 ± 0.000002 eV² this translates to the following D₀(F₂) value:

$$D_0(\text{F}_2) = 15.62294\text{ eV} - 17.42283\text{ eV} + 3.40119\text{ eV} = 1.6012_9 \pm 0.0004_4\text{ eV}$$

This value was further independently confirmed by photoelectron photoion coincidence (PEPICO) experiments which gave the 0 K appearance energy of 19.0242 ± 0.0006 eV for F₂ → F⁺ + F + e⁻, which translates to the following D₀(F₂) value

$$D_0(\text{F}_2) = 19.0242\text{ eV} - 17.42283\text{ eV} = 1.6013_2 \pm 0.0006_2\text{ eV}$$

These two experiment results can then be combined to find a weighted average **F₂ dissociation energy of 1.6013₀ ± 0.0003₆ eV** which is in good agreement to previous ATcT values and improves its accuracy by more than a factor of two.

Acknowledgements

This work was funded by the National Science Foundation (grant no. CHE-1665464). Experiments were performed at the DESIRS VUV beamline of the Soleil Synchrotron under proposal numbers 20180455 and 20190866 and we thank the beamline staff for their support, in particular Dr. Laurent Nahon for helpful discussions in the design and preparation of the experiment. The financial support of the University Research Fund, University of Iceland and the Icelandic Research Fund (Grant No. 184693-051) is gratefully acknowledged. We are grateful to Dr. Krisztián Torma for his help with the cipp and PEPICO experiments and to Dr. Branko Ruscic for fruitful conversations and running our data on top of the latest development version of the Active Thermochemical Tables

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